

TITLE

REFRIGERANT COMPOSITIONS COMPRISING PERFORMANCE
ENHANCING ADDITIVES

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CROSS REFERENCE TO RELATED APPLICATION

This application claims the priority benefit of U.S. Provisional Application 60/402,364, filed August 8, 2002.

FIELD OF THE INVENTION

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The present invention is related to refrigerant and lubricant compositions containing performance-enhancing additives. The performance enhancing additives are found to improve the energy efficiency and/or the capacity of vapor compression refrigeration and air conditioning systems, which utilize hydrofluorocarbon refrigerants and

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oxygenated refrigeration lubricants.

BACKGROUND

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Environmental concerns regarding excessive energy consumption has prompted many industries to modify their commercial products and/or equipment designs to conserve more energy. Consistent with this objective of energy conservation, the refrigeration and air-conditioning industries are constantly seeking to improve energy efficiency of vapor compression refrigeration and air-conditioning systems. For example, additives have been added to the refrigerant/lubricant mixture in

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order to increase the efficiency of vapor compression refrigeration and air-conditioning systems. The additive primarily suggested was tetraethylene glycol dimethyl ether or tetraglyme. However, tetraglyme, and similar glycol ethers with ethylene repeat units have been known to exhibit negative reproductive health effects, which clearly makes them

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undesirable for use in this application.

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Other suggestions have been proposed to address this problem including changes in the design of refrigeration and air-conditioning systems. Unfortunately, many of these suggestions introduce additional complications or disadvantages into the refrigeration and air-conditioning systems.

Accordingly, there is a need for non-toxic, performance-enhancing additives that will improve the energy efficiency and the capacity of vapor-compression refrigeration and air-conditioning systems.

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SUMMARY

The present invention is directed to compositions containing commercially acceptable performance enhancing additives selected from the group consisting of polyoxyalkylene glycol ethers, amides, nitriles, chlorocarbons, aryl ethers 1,1,1-trifluoroalkanes, fluoroethers, lactones, esters, crown compounds, cyclodextrins, and calixarenes. The aforementioned compositions also contain at least one hydrofluorocarbon (HFC), perfluorocarbon (PFC), hydrofluoroether (HFE), ammonia and/or carbon dioxide refrigerant and at least one polyol ester (POE), polyalkylene glycol (PAG), or polyvinyl ether (PVE) lubricants.

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It has been discovered that the use of said additives improves the energy efficiency and/or the capacity of vapor-compression refrigeration and air-conditioning systems. The additives increase the solubility of refrigerants, such as hydrofluorocarbons (HFCs), perfluorocarbons (PFCs), hydrofluoroether (HFE), ammonia and carbon dioxide, in refrigeration lubricants such as polyol esters (POEs), polyalkylene glycols (PAGs), or polyvinyl ethers (PVEs). This increased solubility results in an increase in the heat of vaporization or latent heat. Increased heat of vaporization improves energy efficiency, as well as cooling and heating capacity of vapor compression refrigeration and air-conditioning systems. Additionally, the present additives solubilize marginally soluble fractions, such as low molecular weight oligomers and plasticizers that are typically present and problematic in vapor compression refrigeration systems.

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BRIEF DESCRIPTION OF THE FIGURE

These and other features, aspects, and advantages of the present invention will become better understood with regard to the following description, appended claims, and accompanying figure where:

FIG. 1 is a graph of HFC-134a absorption (mass %) versus time for POE 22 (Mobil Oil product Arctic EAL 22), and the compositions: 70 wt % POE 22, 30 wt % 1-octyl pyrrolidin-2-one (OP); and 70 wt % POE

22, 30 wt % dipropylene glycol dimethyl ether (DMM) as described in Example 2.

DETAILED DESCRIPTION

5 The present inventors discovered that using an effective amount of the present additives in vapor-compression refrigeration and air-conditioning equipment improves performance of these systems. When the additive is combined with POE, PAG or PVE lubricants and refrigerants selected from the group consisting of hydrofluorocarbons
10 (HFCs), perfluorocarbons (PFCs), hydrofluoroethers (HFEs), ammonia and carbon dioxide in a refrigeration system, an improvement in energy efficiency and/or capacity occurs. The present additives improve the energy efficiency and cooling capacity of a vapor-compression refrigeration system by increasing the heat of vaporization upon
15 desorption of the refrigerant from the lubricant and additive composition in the evaporator. Without wishing to be bound by theory, it is believed that breaking the attractions between the refrigerant and the polar, functional group-containing additive results in the increase in heat of vaporization. Likewise, the additives improve the heating capacity of such a system by
20 increasing the heat of solution upon absorption of the refrigerant into the lubricant and additive composition in the condenser.

 The present additives also solubilize low molecular weight oligomers, contained in polymers such as Mylar® and other polyesters. The additives also solubilize plasticizers, such as phthalate esters,
25 stearates and other marginally soluble fractions, which can be extracted from elastomers or polymers by the HFC, PFC or HFE refrigerants, ammonia or carbon dioxide refrigerants or by POE, PAG or PVE lubricants. These extracted fractions can be redeposited in heat exchangers and expansion devices and cause loss of heat transfer and
30 pluggage of tubes and control devices. By keeping these oligomers and plasticizers in solution, they do not form harmful deposits, thus increasing reliability, energy efficiency and overall performance of compression refrigeration equipment.

 In the present compositions comprising lubricant and additive, from about 1 to about 99 weight percent, preferably from about 1 to about 35 60 weight percent, most preferably about 1 to about 20 weight percent of the combined lubricant and additive composition is additive. In terms of

weight ratios, in the present compositions comprising lubricant and additive, the weight ratio of lubricant to additive is from about 99:1 to about 1:99, preferably from about 99:1 to about 40:60, most preferably about 99:1 to about 80:20. Additive may be charged to vapor-compression, 5 refrigeration and air-conditioning systems as a composition of additive, HFCs, PFCs, HFEs, ammonia and/or carbon dioxide refrigerant. When charging compression refrigeration and air-conditioning systems, the additive and refrigerant composition will typically contain from about 0.1 to about 20 weight percent, preferably from about 0.1 to about 10 weight 10 percent additive in the combined additive and refrigerant composition. In vapor-compression refrigeration and air-conditioning systems containing the present compositions comprising HFCs, PFCs, HFEs, ammonia and/or carbon dioxide, lubricant and additive, from about 1 to about 70 weight percent, preferably from about 1 to about 60 weight percent of the 15 refrigerant, lubricant and additive composition is lubricant and additive.

Fluorocarbon refrigerants of the present invention contain at least one carbon atom and one fluorine atom. Of particular utility are fluorocarbons having 1-6 carbon atoms containing at least one fluorine atom, optionally containing oxygen atoms, and having a normal boiling 20 point of from -90°C to 80°C. These fluorocarbons may be represented by the general formula $C_xF_{2x+2-y}H_yO_z$, wherein x is 1-6, y is 0-9, and z is 0-2. Fluorocarbon refrigerants thus include hydrofluorocarbons (HFCs), and perfluorocarbons (PFCs), and hydrofluoroethers (HFEs). Preferred of the fluorocarbons are those in which x is 1-6, y is 1-5 and z is 0-1.
25 Fluorocarbons are commercial products available from a number of sources such as E. I. du Pont de Nemours & Co., Fluoroproducts, Wilmington, DE, 19898, USA, and additionally by synthetic processes disclosed in art such as Chemistry of Organic Fluorine Compounds, edited by Milos Hudlicky, published by The MacMillan Company, New York, N.Y., 30 1962. Representative fluorocarbons include: CHF_3 (HFC-23), CH_2F_2 (HFC-32), CH_3F (HFC-41), CF_3CF_3 (PFC-116), CHF_2CF_3 (HFC-125), CHF_2CHF_2 (HFC-134), CH_2FCF_3 (HFC-134a), CHF_2CH_2F (HFC143), CF_3CH_3 (HFC-143a), CHF_2CH_3 (HFC-152a), CH_2FCH_3 (HFC-161), $CF_3CF_2CF_3$ (PFC-218), $CHF_2CF_2CF_3$ (HFC-227ca), CF_3CFHCF_3 (HFC- 35 227ea), $CHF_2CF_2CHF_2$ (HFC-236ca), $CH_2FCF_2CF_3$ (HFC-236cb), CHF_2CHFCF_3 (HFC-236ea), $CF_3CH_2CF_3$ (HFC-236fa), $CH_2FCF_2CHF_2$ (HFC-245ca), $CH_3CF_2CF_3$ (HFC-245cb), $CHF_2CHFCHF_2$ (HFC-245ea),

CH₂FCHFCF₃ (HFC-245eb), CHF₂CH₂CF₃ (HFC-245fa), CH₂FCF₂CH₂F (HFC-254ca), CH₂CF₂CHF₂ (HFC-254cb), CH₂FCHFCHF₂ (HFC-254ea), CH₃CHFCF₃ (HFC-254eb), CHF₂CH₂CHF₂ (HFC-254fa), CH₂FCH₂CF₃ (HFC-254fb), CH₃CF₂CH₃ (HFC-272ca), CH₃CHFCH₂F (HFC-272ea),
5 CH₂FCH₂CH₂F (HFC-272fa), CH₃CH₂CF₂H (HFC-272fb), CH₃CHFCH₃ (HFC-281ea), CH₃CH₂CH₂F (HFC-281fa), cyclo-C4F₈ (PFC-C318), CHF₂CF₂CF₂CF₂H (HFC-338pcc), CF₃CHFCHFCF₂CF₃ (HFC-43-10mee), C₄F₉OCH₃, and C₄F₉OC₂H₅.

The fluorocarbon refrigerants of the present invention may
10 optionally further comprise azeotropic and azeotrope-like fluorocarbon refrigerant compositions, such as, HFC-125/HFC-143a/HFC-134a (known by the ASHRAE designation, R-404A), HFC-32/HFC-125/HFC-134a (known by ASHRAE designations, R-407A, R-407B, and R-407C), HFC-32/HFC-125 (R-410A), and HFC-125/HFC-143a (known by the ASHRAE
15 designation: R-507).

Refrigerants of the present invention further comprise ammonia and/or carbon dioxide. Additionally, refrigerants may comprise mixtures of fluorocarbon refrigerants and ammonia and/or carbon dioxide.

Refrigerants of the present invention may optionally further
20 comprise up to 10 weight percent of dimethyl ether or at least one C₃ to C₅ hydrocarbon, e.g., propane, propylene, cyclopropane, n-butane, i-butane, and n-pentane. Examples of fluorocarbons containing such C₃ to C₅ hydrocarbons are azeotrope-like compositions of HFC-125/HFC-134a/n-butane (known by the ASHRAE designation R-417A).

25 Refrigeration lubricants of the present invention are those which have been designed for use with hydrofluorocarbon refrigerants and are miscible with fluorocarbons of the present invention under compression refrigeration and air-conditioning systems' operating conditions. Such lubricants and their properties are discussed in
30 "Synthetic Lubricants and High-Performance Fluids", R. L. Shubkin, editor, Marcel Dekker, 1993. Typical refrigeration lubricants have viscosity of at least 9 centistokes at 40 degrees Celsius. Refrigeration lubricants include, but are not limited to, polyol esters (POEs), polyalkylene glycols (PAGs), and polyvinyl ethers (PVEs).

35 Additives of the present invention comprise polyoxyalkylene glycol ethers represented by the formula R¹[(OR²)_xOR³]_y, wherein: x is selected from integers from 1-3; y is selected from integers from 1-4; R¹ is

selected from hydrogen and aliphatic hydrocarbon radicals having 1 to 6 carbon atoms and y bonding sites; R² is selected from aliphatic hydrocarbylene radicals having from 3 to 4 carbon atoms; R³ is selected from hydrogen and aliphatic and alicyclic hydrocarbon radicals having from 1 to 6 carbon atoms; at least one of R¹ and R³ is said hydrocarbon radical; and wherein said polyoxyalkylene glycol ethers have a molecular weight of from about 100 to about 300 atomic mass units and a carbon to oxygen ratio of from about 2.3 to about 5.0. In the present invention, polyoxyalkylene glycol ether additives represented by R¹[(OR²)_xOR³]_y: x is preferably 1-2; y is preferably 1; R¹ and R³ are preferably independently selected from hydrogen and aliphatic hydrocarbon radicals having 1 to 4 carbon atoms; R² is preferably selected from aliphatic hydrocarbylene radicals having from 3 or 4 carbon atoms, most preferably 3 carbon atoms; the polyoxyalkylene glycol ether molecular weight is preferably from about 100 to about 250 atomic mass units, most preferably from about 125 to about 250 atomic mass units; and the polyoxyalkylene glycol ether carbon to oxygen ratio is preferably from about 2.5 to 4.0, most preferably from about 2.7 to about 3.5. The R¹ and R³ hydrocarbon radicals having 1 to 6 carbon atoms may be linear, branched or cyclic. Representative R¹ and R³ hydrocarbon radicals include methyl, ethyl, propyl, isopropyl, butyl, isobutyl, sec-butyl, *tert*-butyl, pentyl, isopentyl, neopentyl, *tert*-pentyl, cyclopentyl, and cyclohexyl. Where free hydroxyl radicals on the present polyoxyalkylene glycol ether additives may be incompatible with certain compression refrigeration apparatus materials of construction (e.g. Mylar®), R¹ and R³ are preferably aliphatic hydrocarbon radicals having 1 to 4 carbon atoms, most preferably 1 carbon atom. The R² aliphatic hydrocarbylene radicals having from 3 to 4 carbon atoms form repeating oxyalkylene radicals - (OR²)_x - that include oxypropylene radicals and oxybutylene radicals. The oxyalkylene radical comprising R² in one polyoxyalkylene glycol ether additive molecule may be the same, or one molecule may contain different R² oxyalkylene groups. The present polyoxyalkylene glycol ether additives preferably comprise at least one oxypropylene radical. Where R¹ is an aliphatic or alicyclic hydrocarbon radical having 1 to 6 carbon atoms and y bonding sites, the radical may be linear, branched or cyclic. Representative R¹ aliphatic hydrocarbon radicals having two bonding sites include, for example, an ethylene radical, a propylene radical, a butylene radical, a pentylene radical, a

hexylene radical, a cyclopentylene radical and a cyclohexylene radical. Representative R¹ aliphatic hydrocarbon radicals having three or four bonding sites include residues derived from polyalcohols, such as trimethylolpropane, glycerin, pentaerythritol, 1,2,3-trihydroxycyclohexane and 1,3,5-trihydroxycyclohexane, by removing their hydroxyl radicals.

Representative polyoxyalkylene glycol ether additives include:

CH₃OCH₂CH(CH₃)O(H or CH₃) (propylene glycol methyl (or dimethyl) ether), CH₃O[CH₂CH(CH₃)O]₂(H or CH₃) (dipropylene glycol methyl (or dimethyl) ether), CH₃O[CH₂CH(CH₃)O]₃(H or CH₃) (tripropylene glycol methyl (or dimethyl) ether), C₂H₅OCH₂CH(CH₃)O(H or C₂H₅) (propylene glycol ethyl (or diethyl) ether), C₂H₅O[CH₂CH(CH₃)O]₂(H or C₂H₅) (dipropylene glycol ethyl (or diethyl) ether), C₂H₅O[CH₂CH(CH₃)O]₃(H or C₂H₅) (tripropylene glycol ethyl (or diethyl) ether), C₃H₇OCH₂CH(CH₃)O(H or C₃H₇) (propylene glycol n-propyl (or di-n-propyl) ether),

C₃H₇O[CH₂CH(CH₃)O]₂(H or C₃H₇) (dipropylene glycol n-propyl (or di-n-propyl) ether), C₃H₇O[CH₂CH(CH₃)O]₃(H or C₃H₇) (tripropylene glycol n-propyl (or di-n-propyl) ether), C₄H₉OCH₂CH(CH₃)OH (propylene glycol n-butyl ether), C₄H₉O[CH₂CH(CH₃)O]₂(H or C₄H₉) (dipropylene glycol n-butyl (or di-n-butyl) ether), C₄H₉O[CH₂CH(CH₃)O]₃(H or C₄H₉) (tripropylene glycol n-butyl (or di-n-butyl) ether), (CH₃)₃COCH₂CH(CH₃)OH (propylene glycol t-butyl ether), (CH₃)₃CO[CH₂CH(CH₃)O]₂(H or (CH₃)₃) (dipropylene glycol t-butyl (or di-t-butyl) ether), (CH₃)₃CO[CH₂CH(CH₃)O]₃(H or (CH₃)₃) (tripropylene glycol t-butyl (or di-t-butyl) ether), C₅H₁₁OCH₂CH(CH₃)OH (propylene glycol n-pentyl ether), C₄H₉OCH₂CH(C₂H₅)OH (butylene glycol n-butyl ether), C₄H₉O[CH₂CH(C₂H₅)O]₂H (dibutylene glycol n-butyl ether), trimethylolpropane tri-n-butyl ether (C₂H₅C(CH₂O(CH₂)₃CH₃)₃) and trimethylolpropane di-n-butyl ether (C₂H₅C(CH₂OC(CH₂)₃CH₃)₂CH₂OH).

The polyoxyalkylene glycol ether additives generally have a kinematic viscosity of less than about 5 centistokes at 40°C. For instance, dipropylene glycol dimethyl ether (DMM) has kinematic viscosity of 0.90 centistokes at 40°C and dipropylene glycol methyl ether has viscosity of 2.7 centistokes at 40°C.

Additives of the present invention further comprise amides represented by the formulae R¹CONR²R³ and cyclo-[R⁴CON(R⁵)-], wherein R¹, R², R³ and R⁵ are independently selected from aliphatic and alicyclic hydrocarbon radicals having from 1 to 12 carbon atoms, and at most one aromatic radical having from 6 to 12 carbon atoms; R⁴ is

selected from aliphatic hydrocarbylene radicals having from 3 to 12 carbon atoms; and wherein said amides have a molecular weight of from about 120 to about 300 atomic mass units and a carbon to oxygen ratio of from about 7 to about 20. The molecular weight of said amides is preferably 5 from about 120 to about 250 atomic mass units. The carbon to oxygen ratio in said amides is preferably from about 7 to about 16, and most preferably from about 10 to about 14. R¹, R², R³ and R⁵ may optionally include substituted radicals, that is, radicals containing non-hydrocarbon substituents selected from halogens (e.g., fluorine, chlorine) and alkoxides 10 (e.g. methoxy). R¹, R², R³ and R⁵ may optionally include heteroatom-substituted radicals, that is, radicals which contain the atoms nitrogen (aza-), oxygen (oxa-) or sulfur (thia-) in a radical chain otherwise composed of carbon atoms. In general, no more than three non-hydrocarbon substituents and heteroatoms, and preferably no more than 15 one, will be present for each 10 carbon atoms in R¹⁻³ and R⁵, and the presence of any such non-hydrocarbon substituents and heteroatoms must be considered in applying the aforementioned ratio of carbon to oxygen and molecular weight limitations. Preferred amide additives consist of carbon, hydrogen, nitrogen and oxygen. Representative R¹, R², 20 R³ and R⁵ aliphatic and alicyclic hydrocarbon radicals include methyl, ethyl, propyl, isopropyl, butyl, isobutyl, sec-butyl, *tert*-butyl, pentyl, isopentyl, neopentyl, *tert*-pentyl, cyclopentyl, cyclohexyl, heptyl, octyl, nonyl, decyl, undecyl, dodecyl and their configurational isomers. Representative R¹, R², R³ and R⁵ aromatic radicals include phenyl, 25 cumenyl, mesityl, tolyl, xylyl, benzyl, phenethyl, thienyl, furyl, pyrrolyl and pyridyl. A preferred embodiment of amide additives are those wherein R⁴ in the aforementioned formula cyclo-[R⁴CON(R⁵)-] may be represented by the hydrocarbylene radical (CR⁶R⁷)_n, in other words, the formula: cyclo-[(CR⁶R⁷)_nCON(R⁵)-] wherein: the previously-stated values for (a) ratio of 30 carbon to oxygen and (b) molecular weight apply; n is an integer from 3 to 5; R⁵ is a saturated hydrocarbon radical containing 1 to 12 carbon atoms; R⁶ and R⁷ are independently selected (for each n) by the rules previously offered defining R¹⁻³. In the lactams represented by the formula: cyclo-[(CR⁶R⁷)_nCON(R⁵)-], all R⁶ and R⁷ are preferably hydrogen, or contain a single saturated hydrocarbon radical among the n methylene units, and R⁵ 35 is a saturated hydrocarbon radical containing 3 to 12 carbon atoms. For example, 1-(saturated hydrocarbon radical)-5-methylpyrrolidin-2-ones.

Representative amide additives include: 1-octylpyrrolidin-2-one, 1-decylpyrrolidin-2-one, 1-octyl-5-methylpyrrolidin-2-one, 1-butylcaprolactam, 1-isobutylcaprolactam, 1-cyclohexylpyrrolidin-2-one, 1-cyclohexyl-5-methylpyrrolidin-2-one, 1-butyl-5-methylpiperid-2-one, 1-pentyl-5-methylpiperid-2-one, 1-hexylcaprolactam, 1-hexyl-5-methylpyrrolidin-2-one, 1-heptyl-5-methylpyrrolidin-2-one, 1-nonyl-5-methylpyrrolidin-2-one, 1-undecyl-5-methylpyrrolidin-2-one, 1-dodecyl-5-methylpyrrolidin-2-one, 5-methyl-1-pentylpiperid-2-one, 1,3-dimethylpiperid-2-one, 1-methylcaprolactam, 1-butyl-pyrrolidin-2-one, 1,5-dimethylpiperid-2-one, 1-decyl-5-methylpyrrolidin-2-one, 1-dodecylpyrrolid-2-one, N,N-dibutylformamide and N,N-diisopropylacetamide.

Additives of the present invention further comprise nitriles represented by the formula R^1CN , wherein R^1 is selected from aliphatic, alicyclic or aryl hydrocarbon radicals having from 5 to 12 carbon atoms, and wherein said nitriles have a molecular weight of from about 90 to about 200 atomic mass units and a carbon to nitrogen ratio of from about 6 to about 12. R^1 in said nitrile additives is preferably selected from aliphatic and alicyclic hydrocarbon radicals having 8 to 10 carbon atoms. The molecular weight of said nitrile additives is preferably from about 120 to about 140 atomic mass units. The carbon to nitrogen ratio in said nitrile additives is preferably from about 8 to about 9. R^1 may optionally include substituted hydrocarbon radicals, that is, radicals containing non-hydrocarbon substituents selected from halogens (e.g., fluorine, chlorine) and alkoxides (e.g. methoxy). R^1 may optionally include heteroatom-substituted hydrocarbon radicals, that is, radicals which contain the atoms nitrogen (aza-), oxygen (keto-, oxa-) or sulfur (thia-) in a radical chain otherwise composed of carbon atoms. In general, no more than three non-hydrocarbon substituents and heteroatoms, and preferably no more than one, will be present for each 10 carbon atoms in R^1 , and the presence of any such non-hydrocarbon substituents and heteroatoms must be considered in applying the aforementioned ratio of carbon to nitrogen and molecular weight limitations. Representative R^1 aliphatic, alicyclic and aryl hydrocarbon radicals in the general formula R^1CN include pentyl, isopentyl, neopentyl, *tert*-pentyl, cyclopentyl, cyclohexyl, heptyl, octyl, nonyl, decyl, undecyl, dodecyl and their configurational isomers, as well as phenyl, benzyl, cumenyl, mesityl, tolyl, xylyl and phenethyl.

Representative nitrile additives include: 1-cyanopentane, 2,2-dimethyl-4-cyanopentane, 1-cyanohexane, 1-cyanoheptane, 1-cyanoctane, 2-cyanoctane, 1-canononane, 1-cyanodecane, 2-cyanodecane, 1-cyanoundecane and 1-cyanododecane.

5 Additives of the present invention further comprise chlorocarbons represented by the formula RCl_x , wherein; x is selected from the integers 1 or 2; R is selected from aliphatic and alicyclic hydrocarbon radicals having 1 to 12 carbon atoms; and wherein said chlorocarbons have a molecular weight of from about 100 to about 200
10 atomic mass units and carbon to chlorine ratio from about 2 to about 10. The molecular weight of said chlorocarbon additives is preferably from about 120 to 150 atomic mass units. The carbon to chlorine ratio in said chlorocarbon additives is preferably from about 6 to about 7.
15 Representative R aliphatic and alicyclic hydrocarbon radicals in the general formula RCl_x include methyl, ethyl, propyl, isopropyl, butyl, isobutyl, sec-butyl, *tert*-butyl, pentyl, isopentyl, neopentyl, *tert*-pentyl, cyclopentyl, cyclohexyl, heptyl, octyl, nonyl, decyl, undecyl, dodecyl and their configurational isomers.

20 Representative chlorocarbon additives include: 3-(chloromethyl)pentane, 3-chloro-3-methylpentane, 1-chlorohexane, 1,6-dichlorohexane, 1-chloroheptane, 1-chlorooctane, 1-chlorononane, 1-chlorodecane, and 1,1,1-trichlorodecane.

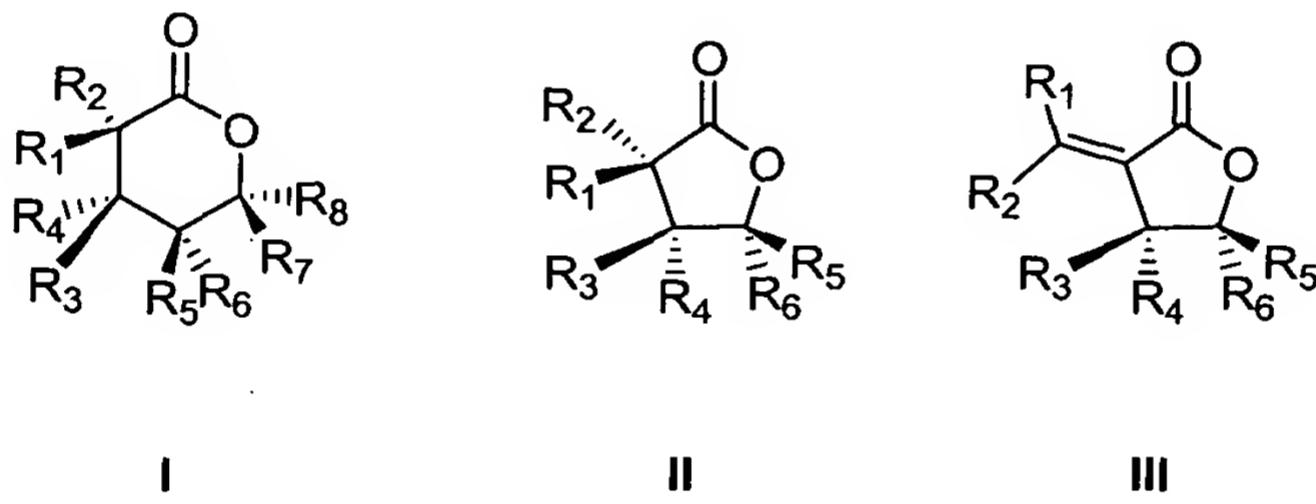
25 Additives of the present invention further comprise aryl ethers represented by the formula R^1OR^2 , wherein: R^1 is selected from aryl hydrocarbon radicals having from 6 to 12 carbon atoms; R^2 is selected from aliphatic hydrocarbon radicals having from 1 to 4 carbon atoms; and wherein said aryl ethers have a molecular weight of from about 100 to about 150 atomic mass units and a carbon to oxygen ratio of from about 4 to about 20. The carbon to oxygen ratio in said aryl ether additives is
30 preferably from about 7 to about 10. Representative R^1 aryl radicals in the general formula R^1OR^2 include phenyl, biphenyl, cumenyl, mesityl, tolyl, xylyl, naphthyl and pyridyl. Representative R^2 aliphatic hydrocarbon radicals in the general formula R^1OR^2 include methyl, ethyl, propyl, isopropyl, butyl, isobutyl, sec-butyl and *tert*-butyl. Representative
35 aromatic ether additives include: methyl phenyl ether (anisole), 1,3-dimethoxybenzene, ethyl phenyl ether and butyl phenyl ether.

Additives of the present invention further comprise 1,1,1-trifluoroalkanes represented by the general formula CF_3R^1 , wherein R^1 is selected from aliphatic and alicyclic hydrocarbon radicals having from about 5 to about 15 carbon atoms, preferably primary, linear, saturated, alkyl radicals. Representative 1,1,1-trifluoroalkane additives include: 1,1,1-trifluorohexane and 1,1,1-trifluorododecane.

Additives of the present invention further comprise fluoroethers represented by the general formula $\text{R}^1\text{OCF}_2\text{CF}_2\text{H}$, wherein R^1 is selected from aliphatic and alicyclic hydrocarbon radicals having from about 5 to about 15 carbon atoms, preferably primary, linear, saturated, alkyl radicals. Representative fluoroether additives include: $\text{C}_8\text{H}_{17}\text{OCF}_2\text{CF}_2\text{H}$ and $\text{C}_6\text{H}_{13}\text{OCF}_2\text{CF}_2\text{H}$.

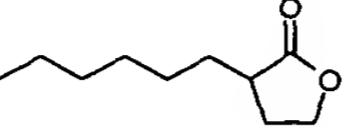
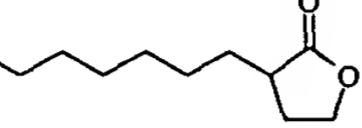
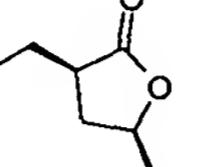
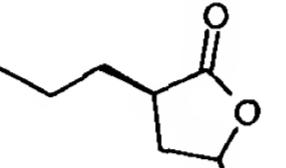
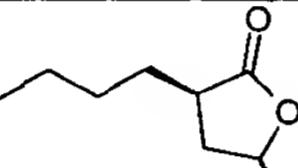
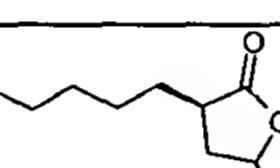
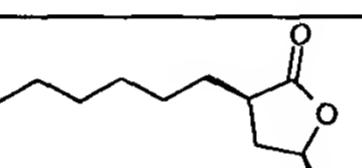
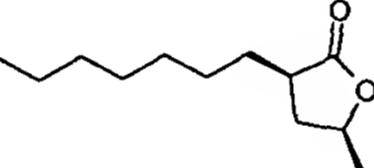
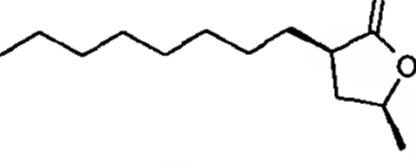
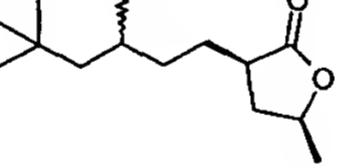
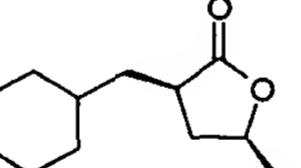
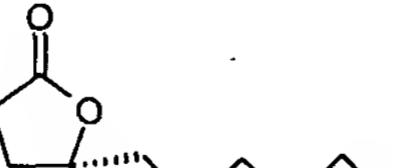
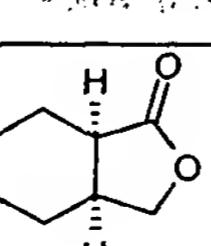
Additives of the present invention further comprise lactones represented by formulas I, II, and III:

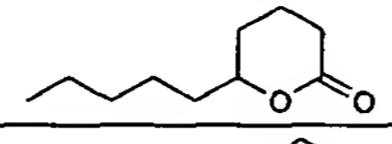
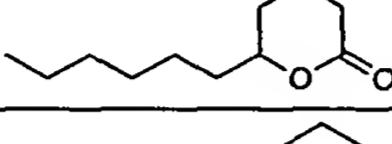
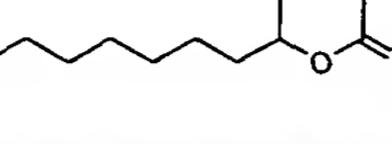
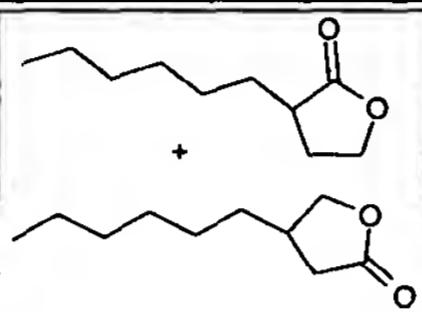
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These lactones contain the functional group $-\text{CO}_2-$ in a ring of six (I), or preferably five atoms (II), wherein for I and II, R_1 through R_8 are independently selected from hydrogen or linear, branched, cyclic, bicyclic, saturated and unsaturated hydrocarbyl radicals. Each R_1 through R_8 may be connected forming a ring with another R_1 through R_8 . The lactone may have an exocyclic alkylidene group as in formula III, wherein R_1 through R_6 are independently selected from hydrogen or linear, branched, cyclic, bicyclic, saturated and unsaturated hydrocarbyl radicals. Each R_1 through R_6 may be connected forming a ring with another R_1 through R_6 . The lactone additives of the present invention have a carbon to ester functional group carbonyl oxygen ratio from about 5 to about 15, preferred from about 5 to about 12. Said lactone additives also have a molecular weight range of from about 80 to about 300 atomic mass units, preferred from about 80 to about 200 atomic mass units. Representative lactone additives include the compounds listed in the table below.

Additive	Molecular Structure	Molecular Formula	Molecular Weight (amu)	Carbon to Carbonyl Oxygen Ratio
(E,Z)-3-ethylidene-5-methyl-dihydro-furan-2-one		C ₇ H ₁₀ O ₂	126	7
(E,Z)-3-propylidene-5-methyl-dihydro-furan-2-one		C ₈ H ₁₂ O ₂	140	8
(E,Z)-3-butylidene-5-methyl-dihydro-furan-2-one		C ₉ H ₁₄ O ₂	154	9
(E,Z)-3-pentylidene-5-methyl-dihydro-furan-2-one		C ₁₀ H ₁₆ O ₂	168	10
(E,Z)-3-Hexylidene-5-methyl-dihydro-furan-2-one		C ₁₁ H ₁₈ O ₂	182	11
(E,Z)-3-Heptylidene-5-methyl-dihydro-furan-2-one		C ₁₂ H ₂₀ O ₂	196	12
(E,Z)-3-octylidene-5-methyl-dihydro-furan-2-one		C ₁₃ H ₂₂ O ₂	210	13
(E,Z)-3-nonylidene-5-methyl-dihydro-furan-2-one		C ₁₄ H ₂₄ O ₂	224	14
(E,Z)-3-decylidene-5-methyl-dihydro-furan-2-one		C ₁₅ H ₂₆ O ₂	238	15
(E,Z)-3-(3,5,5-trimethylhexylidene)-5-methyl-dihydrofuran-2-one		C ₁₄ H ₂₄ O ₂	224	14
(E,Z)-3-cyclohexylmethylidene-5-methyl-dihydrofuran-2-one		C ₁₂ H ₁₈ O ₂	194	12
gamma-valerolactone		C ₅ H ₈ O ₂	100	5
gamma-octalactone		C ₈ H ₁₄ O ₂	142	8
gamma-nonalactone		C ₉ H ₁₆ O ₂	156	9
gamma-decalactone		C ₁₀ H ₁₈ O ₂	170	10
gamma-undecalactone		C ₁₁ H ₂₀ O ₂	184	11
gamma-dodecalactone		C ₁₂ H ₂₂ O ₂	198	12

3-hexyldihydro-furan-2-one		C ₁₀ H ₁₈ O ₂	170	10
3-heptyldihydro-furan-2-one		C ₁₁ H ₂₀ O ₂	184	11
cis-3-ethyl-5-methyl-dihydro-furan-2-one		C ₇ H ₁₂ O ₂	128	7
cis-(3-propyl-5-methyl)-dihydro-furan-2-one		C ₈ H ₁₄ O ₂	142	8
cis-(3-butyl-5-methyl)-dihydro-furan-2-one		C ₉ H ₁₆ O ₂	156	9
cis-(3-pentyl-5-methyl)-dihydro-furan-2-one		C ₁₀ H ₁₈ O ₂	170	10
cis-3-hexyl-5-methyl-dihydro-furan-2-one		C ₁₁ H ₂₀ O ₂	184	11
cis-3-heptyl-5-methyl-dihydro-furan-2-one		C ₁₂ H ₂₂ O ₂	198	12
cis-3-octyl-5-methyl-dihydro-furan-2-one		C ₁₃ H ₂₄ O ₂	212	13
cis-3-(3,5,5-trimethylhexyl)-5-methyl-dihydro-furan-2-one		C ₁₄ H ₂₆ O ₂	226	14
cis-3-cyclohexylmethyl-5-methyl-dihydro-furan-2-one		C ₁₂ H ₂₀ O ₂	196	12
5-methyl-5-hexyl-dihydro-furan-2-one		C ₁₁ H ₂₀ O ₂	184	11
5-methyl-5-octyl-dihydro-furan-2-one		C ₁₃ H ₂₄ O ₂	212	13
Hexahydro-isobenzofuran-1-one		C ₈ H ₁₂ O ₂	140	8

<i>delta</i> -decalactone		C ₁₀ H ₁₈ O ₂	170	10
<i>delta</i> -undecalactone		C ₁₁ H ₂₀ O ₂	184	11
<i>delta</i> -dodecalactone		C ₁₂ H ₂₂ O ₂	198	12
mixture of 4-hexyl-dihydrofuran-2-one and 3-hexyl-dihydro-furan-2-one		C ₁₀ H ₁₈ O ₂	170	10

The lactone additives generally have a kinematic viscosity of less than about 7 centistokes at 40°C. For instance, gamma-undecalactone has kinematic viscosity of 5.4 centistokes and cis-(3-hexyl-5-methyl)dihydrofuran-2-one has viscosity of 4.5 centistokes at 40°C.

Lactone additives may be available commercially or prepared by methods as described in the concurrently filed U. S. provisional patent application, attorney docket no. CL-2362 US PRV, entitled "cis-3,5-Disubstituted-dihydro-furan-2-ones and the Preparation and Use Thereof". The inventors of this provisional application are P. J. Fagan and C. J. Brandenburg. The aforesaid provisional application is incorporated herein by reference.

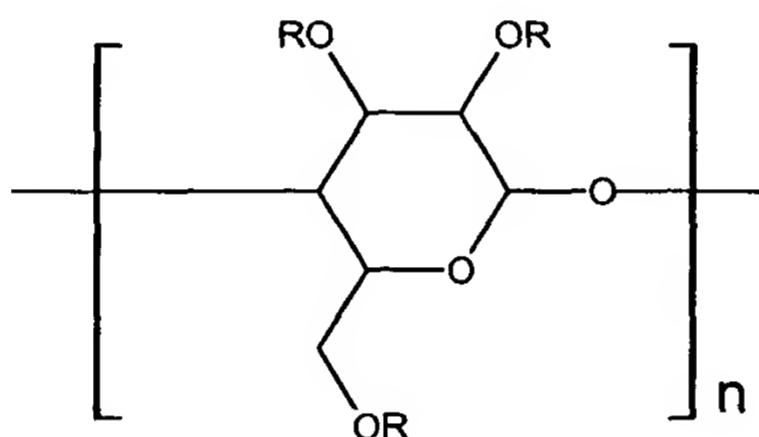
Additives of the present invention further comprise esters represented by the general formula R¹CO₂R², wherein R¹ and R² are independently selected from linear and cyclic, saturated and unsaturated, alkyl and aryl radicals. Preferred esters consist essentially of the elements C, H and O, have a molecular weight of from about 80 to about 550 atomic mass units, and have a carbon to ester functional group carbonyl oxygen ratio of from about 5 to about 15. Representative esters include:

(CH₃)₂CHCH₂OOC(CH₂)₂₋₄OCOCH₂CH(CH₃)₂ (diisobutyl dibasic ester), ethyl hexanoate, ethyl heptanoate, n-butyl propionate, n-propyl propionate, ethyl benzoate, di-n-propyl phthalate, benzoic acid ethoxyethyl ester, dipropyl carbonate, "Exxate 700" (a commercial C₇ alkyl acetate), "Exxate 800" (a commercial C₈ alkyl acetate), dibutyl phthalate, and tert-butyl acetate.

Additives of the present invention further comprise crown compounds with the repeating unit -(CH₂-CH₂-Y)_n- joined in a circular structure, wherein Y is a heteroatom, such as oxygen, nitrogen or sulfur,

and n is greater than 2. Preferred crown compounds are crown ethers wherein Y is oxygen and n is greater than 4. Additionally, the crown compounds may be optionally substituted with side groups containing carbon, hydrogen, oxygen, nitrogen and/or halogen atoms. Crown ethers
5 are available commercially from chemical supply companies such as Aldrich. Representative crown ethers are 18-crown-6 ether, 15-crown-5 ether, and 12-crown-4 ether.

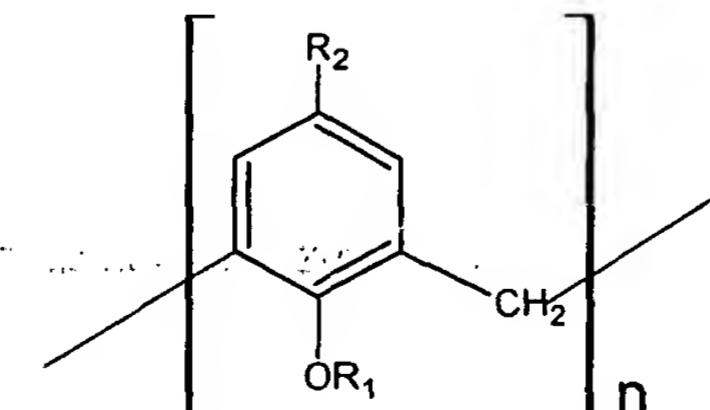
Additives of the present invention further comprise cyclodextrins wherein the repeating units represented by formula IV are
10 connected in a circular structure:



IV

In formula IV, each R group is independently selected from hydrogen or
15 linear, branched, cyclic, bicyclic, saturated and unsaturated hydrocarbyl radicals having up to 10 carbon atoms and n is equal to 6 (alpha-cyclodextrin), 7 (beta-cyclodextrin) or 8 (gamma-cyclodextrin). As an example of a substituted cyclodextrin, perhexyl-beta-cyclodextrin is a cyclodextrin compound comprising 7 repeating units (n = 7) and wherein
20 all R groups comprise hexyl radicals.

Additives of the present invention further comprise calixarenes wherein the repeating units represented by formula V are connected in a circular structure:



In formula V, each R₁ and R₂ group is independently selected from hydrogen or linear, branched, cyclic, bicyclic, saturated and unsaturated hydrocarbyl radicals having up to 10 carbon atoms and n = 4, 5, 6, 7, or 8.

5 Representative calixarenes are calix[4]arene, calix[6]arene, and para-tert-butylcalix[4]arene, wherein n = 4, R₁ comprises hydrogen, and R₂ comprises a tert-butyl radical.

Additives of the present invention may comprise a single additive species or multiple additive species together in any proportion.

10 For example, an additive may comprise a mixture of compounds from within a single additive species (e.g. a mixture of polyoxyalkylene glycol ethers) or a mixture of compounds chosen from different additive species (e.g. a mixture of a polyoxyalkylene glycol ether with an amide).

15 Additives of the present invention may optionally further comprise ketones represented by the formula R¹COR², wherein R¹ and R² are independently selected from aliphatic, alicyclic and aryl hydrocarbon radicals having from 1 to 12 carbon atoms, and wherein said ketones have a molecular weight of from about 70 to about 300 atomic mass units and a carbon to oxygen ratio of from about 4 to about 13. R¹ and R² in said
20 ketones are preferably independently selected from aliphatic and alicyclic hydrocarbon radicals having 1 to 9 carbon atoms. The molecular weight of said ketones is preferably from about 100 to 200 atomic mass units. The carbon to oxygen ratio in said ketones is preferably from about 7 to about 10. R¹ and R² may together form a hydrocarbylene radical
25 connected and forming a five, six, or seven-membered ring cyclic ketone, for example, cyclopentanone, cyclohexanone, and cycloheptanone. R¹ and R² may optionally include substituted hydrocarbon radicals, that is, radicals containing non-hydrocarbon substituents selected from halogens (e.g., fluorine, chlorine) and alkoxides (e.g. methoxy). R¹ and R² may
30 optionally include heteroatom-substituted hydrocarbon radicals, that is, radicals which contain the atoms nitrogen (aza-), oxygen (keto-, oxa-) or sulfur (thia-) in a radical chain otherwise composed of carbon atoms. In general, no more than three non-hydrocarbon substituents and
35 heteroatoms, and preferably no more than one, will be present for each 10 carbon atoms in R¹ and R², and the presence of any such non-hydrocarbon substituents and heteroatoms must be considered in applying the aforementioned ratio of carbon to oxygen and molecular weight

10 20 30 40 50 60 70 80 90 100 110 120 130 140 150 160 170 180 190 200 210 220 230 240 250 260 270 280 290 300 310 320 330 340 350 360 370 380 390 400 410 420 430 440 450 460 470 480 490 500 510 520 530 540 550 560 570 580 590 600 610 620 630 640 650 660 670 680 690 700 710 720 730 740 750 760 770 780 790 800 810 820 830 840 850 860 870 880 890 900 910 920 930 940 950 960 970 980 990 1000 1010 1020 1030 1040 1050 1060 1070 1080 1090 1100 1110 1120 1130 1140 1150 1160 1170 1180 1190 1200 1210 1220 1230 1240 1250 1260 1270 1280 1290 1300 1310 1320 1330 1340 1350 1360 1370 1380 1390 1400 1410 1420 1430 1440 1450 1460 1470 1480 1490 1500 1510 1520 1530 1540 1550 1560 1570 1580 1590 1600 1610 1620 1630 1640 1650 1660 1670 1680 1690 1700 1710 1720 1730 1740 1750 1760 1770 1780 1790 1800 1810 1820 1830 1840 1850 1860 1870 1880 1890 1900 1910 1920 1930 1940 1950 1960 1970 1980 1990 2000 2010 2020 2030 2040 2050 2060 2070 2080 2090 2100 2110 2120 2130 2140 2150 2160 2170 2180 2190 2200 2210 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limitations. Representative R¹ and R² aliphatic, alicyclic and aryl hydrocarbon radicals in the general formula R¹COR² include methyl, ethyl, propyl, isopropyl, butyl, isobutyl, sec-butyl, *tert*-butyl, pentyl, isopentyl, neopentyl, *tert*-pentyl, cyclopentyl, cyclohexyl, heptyl, octyl, nonyl, decyl, 5 undecyl, dodecyl and their configurational isomers, as well as phenyl, benzyl, cumenyl, mesityl, tolyl, xylyl and phenethyl.

Representative ketone additives include: 2-butanone, 2-pentanone, acetophenone, butyrophenone, hexanophenone, cyclohexanone, cycloheptanone, 2-heptanone, 3-heptanone, 5-methyl-2-hexanone, 2-octanone, 3-octanone, diisobutyl ketone, 4-ethylcyclohexanone, 2-nonenone, 5-nonenone, 2-decanone, 4-decanone, 2-decalone, 2-tridecanone, dihexyl ketone and dicyclohexyl ketone.

Ketone additives may have an objectionable odor, which can be masked by addition of an odor masking agent or fragrance. Typical examples of odor masking agents or fragrances may include Evergreen, Fresh Lemon, Cherry, Cinnamon, Peppermint, Floral or Orange Peel or sold by Intercontinental Fragrance, as well as d-limonene and pinene. Such odor masking agents may be used at concentrations of from about 0.001% to as much as about 15% by weight based on the combined weight of odor masking agent and additive.

Additives of the present invention may optionally further comprise from about 0.5 to about 50 weight percent (based on total amount of additive) of a linear or cyclic aliphatic or aromatic hydrocarbon containing from 5 to 15 carbon atoms. Representative hydrocarbons include pentane, hexane, octane, nonane, decane, Isopar® H (a high purity C₁₁ to C₁₂ iso-paraffinic), Aromatic 150 (a C₉ to C₁₁ aromatic), Aromatic 200 (a C₉ to C₁₅ aromatic) and Naptha 140. All of these hydrocarbons are sold by Exxon Chemical, USA.

Additives of the present invention may optionally further comprise polymeric additives. The polymeric additive may be a random copolymer of fluorinated and non-fluorinated acrylates, wherein the polymer comprises repeating units of at least one monomer represented by the formulae CH₂=C(R¹)CO₂R², CH₂=C(R³)C₆H₄R⁴, and CH₂=C(R⁵)C₆H₄XR⁶, wherein X is oxygen or sulfur; R¹, R³, and R⁵ are independently selected from the group consisting of H and C₁-C₄ alkyl radicals; and R², R⁴, and R⁶ are independently selected from the group consisting of carbon-chain-based radicals containing C, and F, and may

further contain H, Cl, ether oxygen, or sulfur in the form of thioether, sulfoxide, or sulfone groups. Examples of such polymeric additives include those disclosed in U.S. patent no. 6,299,792, such as Zonyl® PHS sold by E. I. du Pont de Nemours & Co., Wilmington, DE, 19898, USA.

5 Zonyl® PHS is a random copolymer made by polymerizing 40 weight percent $\text{CH}_2=\text{C}(\text{CH}_3)\text{CO}_2\text{CH}_2\text{CH}_2(\text{CF}_2\text{CF}_2)_m\text{F}$ (also referred to as Zonyl® fluoromethacrylate or ZFM) wherein m is from 1 to 12, primarily 2 to 8, and 60 weight percent lauryl methacrylate ($\text{CH}_2=\text{C}(\text{CH}_3)\text{CO}_2(\text{CH}_2)_{11}\text{CH}_3$, also referred to as LMA).

10 Additives of the present invention may optionally further contain from about 0.01 to 30 weight percent (based on total amount of additive) of an additive which reduces the surface energy of metallic copper, aluminum, steel, or other metals found in heat exchangers in a way that reduces the adhesion of lubricants to the metal. Examples of metal surface energy reducing additives include those disclosed in WIPO PCT publication WO 96/7721, such as Zonyl® FSA, Zonyl® FSP, Zonyl® FSJ and Zonyl® FS62, all products of E. I. du Pont de Nemours and Co. In practice, by reducing the adhesive forces between the metal and the lubricant (i.e. substituting for a compound more tightly bound to the metal), the lubricant circulates more freely through the heat exchangers and connecting tubing in an air conditioning or refrigeration system, instead of remaining as a layer on the surface of the metal. This allows for the increase of heat transfer to the metal and allows efficient return of lubricant to the compressor.

20 Commonly used refrigeration system additives may optionally be added, as desired, to compositions of the present invention in order to enhance lubricity and system stability. These additives are generally known within the field of refrigeration compressor lubrication, and include anti wear agents, extreme pressure lubricants, corrosion and oxidation inhibitors, metal surface deactivators, free radical scavengers, foaming and antifoam control agents, leak detectants and the like. In general, these additives are present only in small amounts relative to the overall lubricant composition. They are typically used at concentrations of from less than about 0.1 % to as much as about 3 % of each additive. These 30 additives are selected on the basis of the individual system requirements. Some typical examples of such additives may include, but are not limited to, lubrication enhancing additives, such as alkyl or aryl esters of

phosphoric acid and of thiophosphates. These include members of the triaryl phosphate family of EP lubricity additives, such as butylated triphenyl phosphates (BTPP), or other alkylated triaryl phosphate esters, e.g. Syn-0-Ad 8478 from Akzo Chemicals, tricresyl phosphates and related compounds. Additionally, the metal dialkyl dithiophosphates (e.g. zinc dialkyl dithiophosphate or ZDDP, Lubrizol 1375) and other members of this family of chemicals may be used in compositions of the present invention. Other antiwear additives include natural product oils and asymmetrical polyhydroxyl lubrication additives such as Synergol TMS (International Lubricants). Natural or synthetic oil-based lubricity enhancing agents may include products such as Zerol ICE from Nu-Calgon, Inc. Similarly, stabilizers such as anti oxidants, free radical scavengers, and water scavengers may be employed. Compounds in this category can include, but are not limited to, butylated hydroxy toluene (BHT), epoxides, and glycidyl ethers.

The present invention further comprises a method of producing refrigeration in a compression refrigeration system containing a HFC, PFC, HFE, ammonia and/or carbon dioxide refrigerant and a refrigeration lubricant selected from the group consisting of POEs, PAGs, and PVEs. The method comprises the step of contacting said lubricant with said refrigerant in the presence of an effective amount of the present additive in said compression refrigeration system.

The present invention further comprises a method of lubricating a compressor in a compression refrigeration apparatus containing HFC, PFC, HFE, ammonia and/or carbon dioxide refrigerant. The method of lubricating comprises the step of adding to said compressor a lubricant and additive composition of the present invention.

The present invention further comprises a process for producing refrigeration in vapor compression refrigeration or air-conditioning systems comprising condensing the present refrigerant compositions, and thereafter evaporating said compositions in the vicinity of a body to be cooled.

The present invention further comprises a process for producing heat comprising condensing the present refrigerant compositions in the presence of a body to be heated, and thereafter evaporating said compositions.

The present invention further comprises a method for improving the energy efficiency and/or capacity of a compression refrigeration apparatus, which comprises the step of using a refrigerant composition of the present invention in a compression refrigeration apparatus.

5 The present invention further comprises a method for reducing deposits and pluggage in a compression refrigeration and/or air-conditioning apparatus comprising the step of using a composition of the present invention in said apparatus.

10 The present invention further comprises a method for delivering an additive to a compression refrigeration apparatus, comprising the step of adding any of the compositions of the present invention to said apparatus.

15 The compositions of the present invention may further be used as cleaning agents, expansion agents for polyolefins and polyurethanes, aerosol propellants, heat transfer media, gaseous dielectrics, fire extinguishing agents, power cycle working fluids, polymerization media, particulate removal fluids, carrier fluids, buffering abrasive agents or displacement drying agents.

20 EXAMPLES

“POE 22” is used herein as an abbreviation for Mobil Oil product Arctic EAL22, a polyol ester lubricant having a kinematic viscosity of 22 centistokes at 40°C. “DMM” used herein is an abbreviation for dipropylene glycol dimethyl ether ($\text{CH}_3\text{O}[\text{CH}_2\text{CH}(\text{CH}_3)\text{O}]_2\text{CH}_3$). “OP” used herein is an abbreviation for 1-octyl pyrrolidin-2-one ($\text{C}_{12}\text{H}_{23}\text{NO}$). “HDO” used herein is an abbreviation for a 1:1 mixture of 4-hexyl-dihydrofuran-2-one and 3-hexyl-dihydrofuran-2-one. DMM, OP, *gamma*-valerolactone, *gamma*-octalactone, *gamma*-nonalactone, *gamma*-decalactone, *gamma*-undecalactone, tetraglyme, 1-methylcaprolactam, 18-crown-6 ether, tert-butyl acetate, N-butylpropionate, N,N-dibutylformamide, and 2-octanone are available from Aldrich Chemical Company (Milwaukee, WI). The synthesis of 1-butylcaprolactam may be carried out as described in the art.

EXAMPLE 1

35 Tests were conducted to measure temperature rise during mixing of HFC and additives of the present invention. Equal volumes of additives and HFC-43-10mee ($\text{CF}_3\text{CF}_2\text{CHFCHFCF}_3$) were mixed in an

insulated Dewar flask outfitted with a temperature probe. Temperature rise was recorded 5 seconds after mixing. Results are shown below.

EXAMPLE 1 – Temperature Rise

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	Initial T(°C)	Final T(°C)	Temp Rise (°C)
Comparative Example			
tetraglyme	21.7	35	13.9
Examples			
1-methylcaprolactam	22.15	35.7	13.55
1-butylcaprolactam	21.1	32.45	11.35
18-Crown-6 ether	22.45	33.2	10.75
n-octyl pyrrolidin-2-one (OP)	20.1	30.75	10.65
Dipropylene glycol dimethyl ether (DMM)	20.7	29.65	8.95
gamma-valerolactone	21.35	30.2	8.85
Tert-butyl acetate	19.7	28.2	8.5
2-heptanone	20.75	28.15	7.4
gamma-octalactone	21	28.3	7.3
N,N-dibutylformamide	22.25	29.1	6.85
N-butylpropionate	21	27.7	6.7
2-nonanone	21.00	27.60	6.6
gamma-nonalactone	21.1	27.5	6.4
gamma-undecalactone (GUDL)	19.7	25.8	6.1
HDFO	20.00	25.95	5.95
2-octanone	19.75	25.5	5.75
gamma-decalactone	21.15	26.45	5.3

Results indicate compositions of the present invention have significant temperature rise indicating high heat of mixing.

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EXAMPLE 2

A refrigerant cylinder and a flat bottom vessel are connected by a short manifold including isolation valves, a fill port, and a pressure transducer. The volume of the entire assembly is pre-determined by mass of water to fill. A known volume of POE 22 lubricant with or without additive (OP or DMM) is charged into the flat bottom vessel at 22°C. Air is evacuated from the headspace. The isolation valve between the two cylinders is closed. HFC-134a refrigerant is loaded into the other cylinder up to a pressure of 181 kPa. The pressure is continuously monitored as the isolation valve between the two cylinders is opened. Over a period of time, the pressure in the system decreases as the refrigerant gas is absorbed into the lubricant phase. An equation of state was used to

translate the reduced pressure into mass percent of refrigerant absorbed into the lubricant/additive mixture. Results are shown in Figure 1. The results indicate that HFC-134a absorption into POE 22 is significantly improved through the addition of additives of the present invention. This
 5 result is indicative of an improvement in heat of vaporization.

EXAMPLE 3

Tests were conducted to determine if the additive could improve performance of R407C (23 wt% HFC-32 and 25 wt% HFC-125,
 10 52 wt% HFC-134a) and polyol ester (POE) oil. A ductless split R22 Sanyo heat pump (evaporator Model KHSO951, condenser Model CHO951) was installed in an environmental chamber. The heat pump was outfitted with an R22 Sanyo rotary compressor (C-1R75H2R). The fan-coil unit was installed in the indoor room of an environmental chamber and the outdoor
 15 unit was installed in the outdoor room. The system was charged with about 1200 grams of R407C and 350 ml of POE oil with a viscosity at 40°C of 46 centistokes. Tests were conducted at ASHRAE cooling B conditions where the indoor room was controlled at 80°F and 67°F wet bulb temperature, the outdoor room at 82°F and 65°F wet bulb temperature.
 20 Tests were also conducted at ASHRAE heating conditions where the indoor room was controlled at 70°F and 60°F wet bulb temperature, the outdoor room at 17°F and 15°F wet bulb temperature relative humidity. Air side capacity, energy efficiency ratio (EER) measurements and oil volume measurements were made. After the
 25 R407C/POE baseline, the oil charge was removed and replaced with additive in POE RL68H so that final oil viscosity at 40°C was comparable to the baseline POE oil. Again, cooling B measurements and low temperature heating measurements were made. Capacity and energy efficiency results are shown below.

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EXAMPLE 3 - Cooling B Test

<u>Oil Composition</u>	<u>Viscosity at 40°C (cs)</u>	<u>Capacity (Kbtu/H)</u>	<u>EER</u>
R407C/50 wt% POE	46	5.91	7.59
Emkarate RL32CF/50 wt%			
POE RL 68H			

R407C/10% tetraglyme in POE RL68H	37	5.68	7.41
R407C/10 wt% n-octyl pyrrolidin-2-one in POE RL68H	46	6.51	8.23

EXAMPLE 3 - Low Temperature Heating Test

<u>Oil Composition</u>	<u>Viscosity at 40°C (cs)</u>	<u>Capacity (Kbtu/H)</u>	<u>EER</u>
R407C/50 wt% POE Emkarate RL32CF/50 wt% POE RL 68H	46	3.71	4.57
R407C/10% tetraglyme in POE RL68H	37	2.92	3.66
R407C/10 wt% n-octyl pyrrolidin-2-one in POE RL68H	46	4.01	4.81

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Results show significantly increased energy efficiency and capacity when n-octyl-pyrrolidin-2-one is added to R407C/POE. Performance is surprisingly improved versus tetraglyme.

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EXAMPLE 4

Tests were conducted to determine if HFC-134a/POE performance could be improved in a domestic refrigerator (Frigidaire 21 cubic foot, Model FRT21P5) using additives of the present invention. The refrigerator was outfitted with pressure and temperature measuring devices as well as power measurement to the hermetic reciprocating compressor and two fans. The compressor was also fitted with a sight glass to monitor lubricant level during operation. The refrigerator was tested in a room controlled at 32.2°C. The refrigerated and freezer compartments were tested at two conditions, mid-mid and warm-warm.

15 The compartment temperatures were monitored and the integrated daily power requirement was calculated. The system was thoroughly flushed between runs to remove residual lubricant. Results are shown below.

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EXAMPLE 4 – Mid-Mid Compartment Temperature Conditions

Lubricant Composition	Oil viscosity at 40°C (cs)	Integrated daily power requirement (kWhr/day)	Freezer compartment temperature (°F)	Refrigerated compartment temperature (°F)
R134a/POE 10	10.0	1.49	3.9	37.4
R134a/10% tetraglyme in POE 10	9.0	1.49	5.1	37.4
R134a/15% n-octyl pyrrolidin-2-one in POE 10	9.3	1.47	3.8	37.6

EXAMPLE 4 – Warm-Warm Compartment Temperature Conditions

Lubricant Composition	Oil viscosity at 40°C (cs)	Integrated daily power requirement (kWhr/day)	Freezer compartment temperature (°F)	Refrigerated compartment temperature (°F)
R134a/POE 10	10.0	1.26	9.6	44.5
R134a/10% tetraglyme in POE 10	9.0	1.26	11.0	44.7
R134a/15% n-octyl pyrrolidin-2-one in POE 10	9.3	1.21	10.0	44.5

Results show an improvement in energy efficiency when n-octyl pyrrolidin-2-one is added to R134a/POE 10. N-octyl pyrrolidin-2-one performance is also surprisingly improved versus tetraglyme.

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